metal-organic papers

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Key indicators

Single-crystal X-ray study T = 140 K Mean σ (S–C) = 0.004 Å Disorder in main residue R factor = 0.029 wR factor = 0.077 Data-to-parameter ratio = 20.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

µ-Oxo-bis[trichlorobis(dimethylsulfane)titanium(IV)]

The structure of $[Ti_2Cl_6O(Me_2S)_4]$ shows a distorted octahedral arrangement around the Ti atom. The molecule does not possess an inversion centre at the O atom. Received 21 July 2003 Accepted 23 July 2003 Online 31 July 2003

Comment

Thio- and selenoether adducts, as well as alkanethiol adducts of titanium tetrachloride have been used as single-source precursors for chemical vapour deposition (CVD) of TiS₂ (Winter *et al.*, 1993) and TiSe₂ (McKarns *et al.*, 1997) layers. The quality of these films was found to be as high as those obtained from the corresponding two-component processes. Along with the CVD studies, the properties of thio and seleno ethers have been investigated (Levason *et al.*, 2000; Wilson *et al.*, 2003; Zeltner *et al.*, 1999). The X-ray structure determination of $[(Me_2S)_2Cl_3Ti]_2(\mu-O), (I)$, is reported here.



Compound (I) crystallizes in the orthorhombic space group $Pna2_1$ with four molecules per unit cell. The titanium metal centres exhibit distorted octahedral arrangements. The thioether ligands are mutually cis, with one trans to the bridging O atom. The chloride and thioether ligands bend away from the oxo ligand $[O-Ti-Cl = 95.71 (8)-99.64 (8)^{\circ}; O-Ti-S =$ 93.69 (8) and 96.64 (8)°]. An almost identical arrangement can be found in $[(MeS(CH_2)_2SMe)Cl_3Ti]_2(\mu-O)$ (Levason *et al.*, 2000); here an inversion centre situated at the bridging O ligand links the two distorted octahedral titanium(IV) centres. However, compound (I) does not possess an inversion centre [Ti-O = 1.778(2) and 1.812(2) Å, and Ti-O-Ti =171.86 (14)°]. Related oxo-bridged homodinuclear titanium species are known, both with $\{[(\eta^5 C_5 H_5) Ti Cl_2]_2(\mu - O);$ Thewalt & Schomburg, 1977} and without {[THF₂Cl₃Ti]₂(μ -O); Streltsova et al., 1988; Mahrwald et al., 2001} an inversion centre in the oxo bridge.

Experimental

Compound (I) was formed from the hydrolysis product $Cl_3Ti(\mu-O)TiCl_3$ contaminating the starting material $TiCl_4$; a thioether adduct was formed. The title compound was isolated after crystallization from petroleum ether at 253 K as orange blocks.

© 2003 International Union of Crystallography Printed in Great Britain – all rights reserved Crystal data

 $\begin{array}{l} [\text{Ti}_2\text{Cl}_6\text{O}(\text{C}_2\text{H}_6\text{S})_4] \\ M_r = 573.01 \\ \text{Orthorhombic, } Pna2_1 \\ a = 25.311 \ (5) \text{ Å} \\ b = 9.801 \ (2) \text{ Å} \\ c = 9.4720 \ (19) \text{ Å} \\ V = 2349.7 \ (8) \text{ Å}^3 \\ Z = 4 \\ D_x = 1.620 \ \text{Mg m}^{-3} \end{array}$

Data collection

Rigaku R-AXIS-IIc diffractometer φ scans Absorption correction: none 11796 measured reflections 4350 independent reflections

Refinement

Refinement on F^2
$R[F^2 > 2\sigma(F^2)] = 0.029$
$wR(F^2) = 0.077$
S = 1.06
4153 reflections
201 parameters
H-atom parameters constrained

Mo $K\alpha$ radiation Cell parameters from 221 reflections $\theta = 1.6-22.7^{\circ}$ $\mu = 1.71 \text{ mm}^{-1}$ T = 140 (1) KBlock, orange $0.40 \times 0.35 \times 0.30 \text{ mm}$

4021 reflections with $I > 2\sigma(I)$ $\theta_{\text{max}} = 25.4^{\circ}$ $h = -30 \rightarrow 30$ $k = -11 \rightarrow 11$ $l = -11 \rightarrow 11$

 $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0544P)^{2} + 0.4789P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 0.29 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{min} = -0.46 \text{ e} \text{ Å}^{-3}$ Absolute structure: Flack (1983), 1846 Friedel pairs Flack parameter = 0.45 (3)

The disordered methyl group C42 was modelled in two positions and their ratio of occupancy was refined to 59:41 (7). H atoms were included in idealized positions (C–H 0.960 Å) and their $U_{\rm iso}$ values were set at 1.5 times $U_{\rm eq}$ of the parent C atom. Compound (I) crystallizes as a twin, and the Flack parameter represents the contribution of the second polar axis component.

Data collection: *R-AXIS Control Software* (Molecular Structure Corporation, 1994); cell refinement: *DENZO* (Otwinowski & Minor, 1997); data reduction: *DENZO* and SCALEPACK (Otwinowski & Minor, 1997); program(s) used to solve structure: *XS* in *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *XL* in *SHELXTL*; molecular graphics: *XP* in *SHELXTL*; software used to prepare material for publication: *XL* in *SHELXTL*.

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Figure 1

A view of (I), with displacement ellipsoids drawn at the 40% probability level. Hydrogen atoms have been omitted. Both disorder components are shown.

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